

JP 51-046581

At page 4, the upper right column, line 16 to the lower right column, line 10.

The above-described melting-ejection enables a high melting point material also to be a target of coating by melting because the melting-ejection is a high temperature operation. Further, the melting-ejection can form a coat of an insoluble anode showing a good adhesion as well as metals, their compounds (generally, oxide), or their original minerals, more than one kind of element and combinations of their oxides can be used as a coding material for the insoluble anode for formation. For example, oxides, elements in which oxygen is removed, and their combinations become targets and can be used, in which the oxides are such as (to be continued)

Pb_2O , SiO , GeO , Cu_4O ,
 Cd_2O , SrO_2 , Os_2O_2 , TiO_3 , CuO_2 , Th_2O_7 ,
 Rb_2O_2 , ReO_4 , As_2O_5 , ZnO , $\text{Al}_2\text{O}_3 \cdot x\text{H}_2\text{O}$,
 Sb_2O_3 , $\text{Sb}_2\text{O}_5 \cdot x\text{H}_2\text{O}$, SbO_2 , Yb_2O_3 , Y_2O_3 ,
 Ir_2O_3 , IrO_2 , In_2O , InO , In_2O_3 , UO_2 ,
 $\text{UO}_3 \cdot x\text{H}_2\text{O}$, U_3O_8 , Er_2O_3 , CeO , Os_2O_3 , OsO_2 ,
 OsO_4 , CdO , Cd_2O_3 , Ga_2O , GaO , Ga_2O_3 ,
 CaO , Au_2O , Au_2O_3 , As_2O , AgO , CrO ,
 Cr_2O_3 , CoO , Co_2O_3 , Co_3O_4 , Sm_2O_3 , Sm_4O_9 ,
 Dy_2O_3 , ZrO_2 , Hg_2O , HgO , AuO(OH) ,
 $\text{Si}_2\text{O}_2(\text{OH})_2$, $\text{Pb}_3\text{O}_3(\text{OH})_2$, $\text{Ni}_3\text{O}_2(\text{OH})_4$,
 MnO(OH) , MnO(OH)_x , Se_2O_5 , SnC , $\text{SnO}_2 \cdot x\text{H}_2\text{O}$,
 SrO , Os_2O , Ce_2O_3 , OsO_2 , Ta_2O , Ta_2O_3 ,
 WO_2 , $\text{WO}_3 \cdot x\text{H}_2\text{O}$, TaO , Ta_2O_5 , TlO , Ti_2O_3 ,
 TiO_2 , $\text{TiO}_2 \cdot x\text{H}_2\text{O}$, FeO , Fe_2O_3 , Fe_3O_4 ,
 Tb_2O_3 , Tb_4O_7 , TeO , Cu_2O , CuC , ThO_2 , PbO ,
 PbO_2 , Pb_2O_3 , Pb_3O_4 , NbO , NbO_2 , Nb_2O_5 ,
 HfO , Hf_2O_3 , Hf_3O_4 , Nd_2O_3 , NpO_2 , PtO .

PtO_2 , Pt_3O_4 , VO , V_2O_3 , VO_2 , $\text{V}_2\text{O}_5 \cdot x\text{H}_2\text{O}$,

HfO_2 , Pd_2O , PdO , PdO_2 , BaO , Bi_2O_3 ,

$\text{Bi}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$, $\text{Bi}_2\text{O}_5 \cdot x\text{H}_2\text{O}$, Pr_2O_3 , PrO_2 ,

PuO_2 , BeO , $\text{B}_2\text{O}_3 \cdot x\text{H}_2\text{O}$, MgO , MnO , Mn_2O_3 ,

$\text{MnO}_2 \cdot x\text{H}_2\text{O}$, MnO_3 , Mn_2O_7 , Mn_3O_4 , MoO_2 ,

$\text{MoO}_3 \cdot x\text{H}_2\text{O}$, Eu_2O_3 , La_2O_3 , Li_2O , Ru_2O_3 ,

RuO_2 , Ru_2O_5 , RuO_4 , Rb_2O , ReO_2 , Re_2O_7 ,

RhO , Rh_2O_3 , RhO_2 , TeO_3 , B_2O_3 , Rb_2O_3 .

BEST AVAILABLE COPY